ENRP 6101: Environmental Sciences I: Physical Sciences Fall 2018

Time and Location	Mondays, 6:10 – 8:40 pm; COR 104
Instructor	Dr. Velmurugan "Vel" Subramanian, PE, BCEE, PMP, LEED AP Phone: 918 504 4152 Email: svel65@hotmail.com; <u>vels@gwu.edu</u>
Office Hours	By Appointment
Course Materials	Living in the Environment: Concepts, Connections, and Solutions by Miller/Spoolman, 16 th Edition
	Environmental Science: Earth as Living Planet by Botkin and Keller, 8 th Edition
	The Earth System by Kump, Kasting and Crane 3 rd Edition
Course Catalog Description	This course explores basic physical sciences crucial to environmental issues, including chemistry, geology, hydrology, climate science, and cross-media interactions; land, air, and water pollution, climate change, production and consumption of energy, sea level rise, and anthropogenic changes in the cryosphere. Emphasis is placed on students' gaining a broad and yet significantly deep scientific knowledge to understand the scope and extent of human-environment interactions.
Learning Outcomes	At the completion of this course, students are expected to be able:
	 To become acquainted with the primary scientific disciplines of environmental sciences (i.e. atmospheric science, geoscience, and hydroscience) and understand how principles in these disciplines collectively govern the natural behavior of earth environment. To understand the complexity of environmental problems by learning the multiplicity of causes and consequences of major environmental issues we are facing. To be able to define and explain the basic issues concerning the ability of human community to interact in a sustainable manner with the environment. To understand the importance of 'system approach' in environmental issues To be able to evaluate scientific information and arrive at defendable strategies as to their validity and applicability to improving environment qualities in an economic fashion.

Grading	Homework Exam 1 Exam 2 Current environmental issues Research Paper/Presentation In-Class Problems/Activities	25% 22.5% 22.5% 5% 20% 5%
		+ = 87 – 89.9% B = 83-86.9% B- = 80 – .9% C- = 70 – 72.9% D+ = 67 – 69.9% D = <60%
Homework/Research Paper	date. Hard copies shall be turn	ents are due at start of class on specified ned in. Late homework/projects will not vill miss a class, it is your responsibility to vance.
	term. This assignment will be c a written report and a presenta	ch assignment during the course of the arried out with a partner and will require ation. A detailed description of this nts/expectations will be discussed during
Current Environmental Issues	environmental issue of importa	ke a 5 minute presentation of any ance. Every week there will be two I be have to sign up for the date in
Readings		ompleted prior to the day of class when will be either from the text or from
Academic Integrity	Cheating and plagiarism in any home works are to be individua	emic honesty is expected in the class. of their forms are unacceptable. All al work. The University Code of Academic //www.gwu.edu/~ntegrity/code.html.
Laptop/Cell Phone Policy	activities not directly related to earphones, etc. are to be used rings, musical interludes, etc. a related or family emergency th please inform me of this situati violate these class room policie	ed for web surfing, e-mail, or any other o class. No headphones, blue tooth in the classroom. No cell-phone calls, illowed during class. If you have a work hat requires you to take a call during class, ion in advance of each class period. If you es, you will be asked to leave the class and period in respect of other student's need

Religious Holidays	To be complied with the university policy regarding religious holidays.
Attendance Policy	Students are expected to be present for every meeting of classes. Any anticipated absences must be cleared with the instructor, in advance or the absence will be considered unexcused. A student may have no more than two unexcused absences. Three unexcused absences will result in a grade reduction of one letter grade. Four unexcused absences will result in a grade of "I" in the course.
Disability Support Services	Any student who may need an accommodation based on the potential impact of a disability should contact the Disability Support Services to establish eligibility and to coordinate reasonable accommodations.
Emergency Preparedness	For emergency preparedness information, please refer to: https://smhs.gwu.edu/about/emergency-preparedness-resources

SYLLABUS

WEEK	DATE	TOPICS
		BASICS
Week 1	Aug 27	Course Introduction Some important definitions Environmental issues and scales Environmental sciences Causes of Environmental Problems Environmental supply and human demand Human population growth and associated demand Pollution Case study: Minamata disease Goal of Environmental Science Concept of sustainability Scientific principles for maintaining natural sustainability Earth carrying capacity Environmental History: An Overview
Week 2/3	Sep 10/17	Scientific Principles and Concepts Understanding what science is Observations, Facts and Interferences Controlling variables Nature of scientific proof Theory in science and language Models and Theory Measurements and Uncertainty A word about numbers in science Dealing with uncertainties Accuracy and precision Misunderstanding about Science and Society Science and decision making Science and technology Science and technology Science, pseudoscience and frontier science Environmental Questions and the Scientific Method
Week 4	Sep 24	System Approach, Global Energy Budget and Redistribution System and its components; feedback loops and coupling Behavior of complex systems: time delay/synergy/resistance Black body radiation and earth surface temperature Global energy budget and greenhouse effect Global energy redistribution: earth surface fluid circulation
	0.1.4	LITHOSPHERE
Week 5	Oct 1	Past, Current, and Future of the Earth Origin and evolution of the earth and the solar system Physical and Chemical Structure of the solid Earth

		Earth Materials and Mineral/Rock Resources Rocks and Minerals classified by their compositions Elements' crustal abundance and concentration factor for ores Common economical minerals and their uses Mineral and rock resource and related environmental issues
Week 6	Oct 15	Energy Resources Nonrenewable vs. renewable resources Geological process of fossil fuel formation Nuclear energy and nuclear waste Hydropower, geothermal, solar, and wind energy Environmental problems related to energy resource exploration
Week 7	Oct 22	Soil Science Weathering and soil formation; soil mineralogy Soil taxonomy and profile Soil chemistry: colloids, ion adsorption/exchange, alkalinity/salinity Soil water and hydrology Nutrient cycling and fertility
		ATMOSPHERE
Week 8	Oct 29	Atmospheric Science Basic composition Ideal Gas Law The structure of the atmosphere Basic atmospheric energetic: Pressure and Temperature Basic tropospheric and stratospheric and chemistry Ozone chemistry Chemistry of air pollution
Week 9	Nov 5	Air Pollution and Consequences-I: Greenhouse Gases and Global Climate Anthropogenic input of GHGs Evidence of anthropogenic CO2 buildup since industrial revolution Past climate and historical CO2 level Projected future CO2 level and associated T and sea level change Carbon reservoirs and fluxes Natural regulation of atmospheric CO2: C cycles
Week 10	Nov 12	 Air Pollution and Consequences-II: CFCs and Ozone Depletion Ozone in the atmosphere and the UV effect Mechanisms for CFC destroying ozone: Antarctic ozone hole Montreal Protocol and current state of CFC substitution Air Pollution and Consequences-III: SO₂, NOx and Acid Precipitation Natural and anthropogenic sources of SO2, NOx Chemistry of acid precipitation formation Clean Air Act

		HYDROSPHERE
Week 11	Nov 19	Water Resources and Hydrology Global water cycle: surface vs. groundwater Surface water: erosion and flooding Groundwater: Darcy's law Surface and groundwater interaction: salt water intrusion Hydrofracking: Groundwater contamination? Clean Water Act
Week 12	Nov 26	Interactions of Water with Atmosphere and Lithosphere Regulation of water pH by CO ₂ -H ₂ O equilibria Measure of water quality Control of water chemistry by mineral dissolution/precipitation Acid mine drainage
Week 13	Dec 3	Water Quality Water quality parameters and its measurement Waterborne disease Nutrients Water pollution Wastewater treatment Water reuse
	Dec 10	EXAM 2/PROJECT PRESENTATION